RIGIDIN, A NOVEL ALKALOID WITH CALMODULIN ANTAGONISTIC ACTIVITY FROM THE OKINAWAN MARINE TUNICATE EUDISTOMA CF. RIGIDA

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Summary: A novel pyrrolopyrimidine alkaloid, rigidin (1), with calmodulin antagonistic activity has been isolated from the Okinawan marine tunicate Eudistoma cf. rigida. The structure was elucidated on the basis of spectral data of 1 and its pentamethyl derivative (2).

In our continuing search for bioactive compounds from Okinawan marine organisms, we have encountered a purple-colored compound tunicate *Eudistoma* cf. *rigida* to result in the isolation of the two novel cytotoxic 24-membered macrolides, iejimalides A and B.² In this communication we wish to report the isolation and structure elucidation of a novel pyrrolopyrimidine alkaloid, named rigidin (1), with potent calmodulin antagonistic activity from the same tunicate.

The tunicate was collected at Ie Island, Okinawa, by SCUBA (-5 ~ -15 m) and kept frozen until used. The methanol extract was partitioned between toluene and water. The aqueous layer was subsequently extracted with chloroform, ethyl acetate, and n-butanol. The ethyl acetate soluble material was subjected to a silica gel column (CHCl₃/n-BuOH/AcOH/H₂O, 3:6:1:1) followed by a Sephadex LH-20 column (CHCl₃/MeOH, 1:1) to give rigidin (1, 0.0015% wet weight) as a purple solid: mp >300 °C.

The molecular formula, $C_{19}H_{13}N_3O_5$, of 1 was established by HRFABMS (m/z 364.0917, M+H⁺, Δ -1.6 mmu). Elemental analysis suggested that rigidin (1) was isolated as a free base. The UV absorption indicated the presence of phenol chromophore(s).³ The IR spectrum was indicative of the presence of hydroxyl and unsaturated carbonyl groups.³ The ¹H NMR spectrum of 1 showed five NH/OH protons (δ 9.27 ~ 11.78 in DMSO- d_6) and eight aromatic protons in two A₂B₂ systems (δ 7.29, 2H, J=8.6 Hz; 6.94, 2H, J=8.6 Hz; 6.44, 2H, J=8.6 Hz). The ¹H-¹H COSY spectrum showed a weak but

Table 1. ^{1}H and ^{13}C NMR data for pentamethyl rigidin (2) in CDCl₃

Position	δ_{C} (m)	δ _H (m)	J (Hz)	H coupled with C
1	150 40 (-)			11 22 11 24
2 3	152.40 (s)			H ₃ -23, H ₃ -24
3	150 (7 (-)			11 22
4	158.67 (s)			H_3-23
4a	99.57 (s)			77 0 10
5	129.23 (s)			$H_2-9,13$
6	129.63 (s)			H ₃ -25
7				
7a	142.81 (s)			H ₃ -24, H ₃ -25
8	123.87 (s)			$H_2-10,12$
9,13	132.12 (d)	7.09 (d)	8.8	H_2 -10,12
10,13	113.16 (d)	6.61 (d)	8.8	H ₂ -9,13
11	158.94 (s)			H ₂ -9,13, H ₃ -22
14	186.79 (s)			H ₂ -16,20
15	130.72 (s)			$H_2-17,19$
16,20	132.23 (d)	7.55 (d)	8.8	- ,
17,19	112.83 (d)	6.59 (d)	8.8	
18	163.12 (s)	0.05 (0)		H ₂ -16,20, H ₃ -21
21	55.38 (q)	3.74 (s)		
22	55.16 (q)	3.74 (s)		
23	28.29 (q)	3.38 (s)		
24	33.92 (q)	3.85 (s)		
25	35.78 (q)	3.94 (s)		

clear cross peak between the protons at δ 11.21 (H-23 or H-24) and 10.64 (H-24 or H-23), which may be assigned to a W-coulpling. Since no further structural information was available from the NMR data of 1 itself, the structure determination was mostly carried out with its pentamethyl derivative (2). 2 was obtained as major product on methylation of 1 with CH₂N₂ in methanol. The ¹H NMR of 2 (Table 1) showed no longer NH/OH protons, but five singlet methyls at $3 \sim 4$ ppm region, along with eight A_2B_2 aromatic protons just as in 1. The EIMS of 2 showed the molecular ion at m/z 433 and a sole fragment ion at m/z135. The HREIMS established the molecular formula of 2 as $C_{24}H_{23}N_{3}O_{5}$ (Found m/z433.1610, Δ-2.8 mmu). Among five methyls in the ¹³C NMR of 2 (Table 1) two resonating at 8 55.16 and 55.38 were assured to methoxyls, and the remaining three were all assigned to N-methyls due to their higher-field resonances (δ 28.29, 33.92, and 35.78). Long-range ¹H-¹³C couplings observed by HMBC⁴ experiments (Table 1) suggested the presence of two p-methoxyphenyl moieties by the following correlations: H₃-22/C-11, H₂-10,12/C-8, H₂-9,13/C-11; H₃-21/C-18, H₂-16,20/C-18 and H₂-17,19/C-15. This was also confirmed by NOE experiments. Irradiation of the H₃-22 signal resulted in 5.4% NOE for H₂-10,12, while irradiation of the H₃-21 signal caused 6.1% NOE for H₂-17,19. In the HMBC spectrum the H_2 -9,13 also coupled with an sp² carbon at δ 129.23 (C-5), which showed no coupling to other protons and therefore should be located at four-bond away from any other proton. The H₂-16,20, however, showed a cross peak to the lowest field carbon (δ 186.79), which should be assigned to a cross-conjugated carbonyl group (C-14). The intense fragment ion peak at m/z 135 was ascribable to the methoxybenzoyl substituent $(C_8H_7O_2)$. The observation of 2.1% NOE of H_2 -9,13 on irradiation of H_2 -16,20 strongly suggested the presence of partial structure B. The assignment of partial structure A consisting of N-1 ~ C-4, C-7a and N-7 atoms was straightforward: the methyl at δ 3.38 (H₃-23) coupled to C-4 (\delta 158.67) and C-2 (\delta 152.40), the latter in turn coupled to the methyl at δ 3.85 (H₃-24) in the HMBC spectrum. Similarly, the H₃-24 and H₃-25 (δ 3.94) showed common cross peak to a carbon at δ 142.81 (C-7a).

Partial structures A and B accounted for 23 carbons, and only one carbon and three unsaturated degrees remained to be assigned. Since the carbon at δ 129.63 (C-6) coupled to H₃-25 protons, it should be connected directly to 7-N atom (three bond away from Me protons) and must be C-6. Therefore the remaining carbon (C-4a) has a chemical shift of δ 99.57 and connected to C-7a and C-5 to constitute a pyrrolopyrimidine moiety. Thus the structure of pentamethyl rigidin was concluded to be 2 and that of rigidin itself was assigned to 1.

Rigidin (1) exhibited potent calmodulin antagonistic activity.^{5,6} The value of the 50% inhibitory concentration of calmodulin-activated brain phosphodiesterase was 5 x 10⁻⁵ M. Although pyrrolo[2,3d]pyrimidine type compounds have been found in some nucleoside antibiotics,⁷ isolated from strains of *Streptomyces*, there are no precedents in marine sources.⁸ Interestingly, pyrrolo[2,3d]pyrimidine-2,4-diones have been synthesized and some of them showed weak affinity for the benzodiazepine receptor.⁹

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- 3. UV (MeOH) λ max 401 (ϵ 2500), 346 (2500), 285 (4300), and 240 nm (sh); (MeOH+KOH) λ max 401 (ϵ 2500), 346 (4500), 285 (4300), and 240 nm (sh); (MeOH+HCl) λ max 552 (ϵ 200), 356 (2500), 276 (4500), and 232 (sh) nm; IR (KBr) 3200, 1700, 1610, 1420, and 1270 cm⁻¹.
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